HYDROPEROXYCYCLOARTANES FROM TILLANDSIA RECURVATA

GABRIELA M. CABRERA and ALICIA M. SELDES*

Departamento de Química Orgánica, Facultad de Ciencias Exactas y Naturales, Universidad de Buenos Aires, Pabellon 2, Ciudad Universitaria, 1428 Buenos Aires, Argentina

ABSTRACT.—25-Hydroperoxycycloart-23-en-3 β -ol [1] and an epimeric mixture of 24hydroperoxycycloart-25-en-3 β -ol [2] have been isolated from *Tillandsia recurvata*. The structures of compounds 1 and 2 have been established by spectral analysis and chemical conversion to the corresponding diols.

As part of our continuing interest in plants of the genus Tillandsia and their metabolites (1), we have examined an extract of Tillandsia recurvata (L.) L. (Bromeliaceae), an aerial plant endemic to Argentina. We now report the isolation of two new cycloartane triterpenes with an unusual hydroperoxide functional group, 25-hydroperoxycycloart-23-en-3B-ol [1] and 24-hydroperoxycycloart-25-en-3 β -ol [2], from an extract of the fresh plant. In addition to these new compounds, we isolated and identified the known compounds cycloartanone, cycloartenone, 24-methylenecycloartanone, cycloartanol, cycloartenol, 24-methylenecycloartanol, lanostenol, lanosterol, and 24-ethylcholest-4-en-3-one.

Compound **1** showed a molecular ion (eims, 20 eV) at m/z 458 corresponding to the molecular formula $C_{30}H_{50}O_3$. The ¹H-nmr spectrum displayed signals due to six tertiary [δ 0.81, 0.88, 0.97 (6H), 1.34 (6H)] and one secondary (δ 0.87, d, J=6.5 Hz) methyl groups. A pair of doublets at δ 0.33 and 0.56 (J=4.3

Hz) was indicative of a cyclopropane ring. The double doublet at δ 3.28 (J=10.6and 5.1 Hz) was due to a proton attached to a carbon bearing a hydroxy group. Two trans olefinic protons appeared at δ 5.70 (ddd, J=15.7, 8.1, and 5.9 Hz) and 5.52 (d, J=15.7 Hz), and a broad singlet corresponding to an exchangeable proton appeared at δ 7.32. The ¹³C-nmr spectrum showed signals for 30 carbon atoms. The multiplicity assignments were made by DEPT experiments. Four low-field signals were assigned to a double bond (δ 130.7 and 134.5), an oxygenated methine $(\delta 78.9)$, and an oxygenated quaternary carbon (δ 82.3). The high-field region showed ¹³C-nmr resonances of the ABCD ring carbons similar to those reported for cvcloartenol(2) and additional signals for a methine (δ 36.3), a methylene (δ 39.4), and three methyl carbons (δ 18.4, 24.3, and 24.4).

¹H-¹H COSY and long-range homonuclear COSY spectra exhibited correlations that allowed us to obtain the partial structure shown in Figure 1. Taking into





account the molecular formula, the substituent was identified as a hydroperoxide. Furthermore, a NOESY spectrum showed a nOe correlation between the exchangeable proton at δ 7.32 and H-24. Reduction of **1** with LiAlH₄ afforded **3** which was identified by direct comparison (tlc and ¹H-nmr) with an authentic sample (3,4).

Compound 2 exhibited a molecular ion at m/2 458 (eims, 20 eV). Comparison of the ¹H- and ¹³C-nmr spectra (Tables 1 and 2) with those of compound 1 suggested that compound 2 was also a 3βhydroxycycloartane, isomeric with 1. The most important differences between the ¹H-nmr spectra of 1 and 2 were the absence of the olefinic proton signals for H-23 and H-24 (δ 5.52 and 5.70), the methyl singlet for H-26, 27 (δ 1.34), and the presence of a two-proton broad singlet (δ 5.01), a triplet (δ 4.27, J=6.5 Hz), and a vinylic methyl group (δ 1.74) in the latter compound. The exchangeable proton signal was still present, but shifted downfield to δ 7.78. Differences between the ¹³C-nmr spectra of the two compounds were consistent with a different location of the double bond and the hydroperoxide group in the side-chain. In the 13 C-nmr spectrum of **2**, in particular, all the signals corresponding to the side-chain were duplicated, suggesting that 2 was actually a mixture of two epimers (5). DEPT nmr experiments showed that the double bond was formed between a quaternary carbon and an sp^2 methylene. This fact and the presence of a vinylic methyl group suggested the presence of an isopropenyl group in the side-chain. The ¹H-¹H COSY nmr spectrum of 2 showed correlations between the broad singlet at δ 5.01 and the signals at δ 1.74 and 4.27, thus locating the

Proton(s)	1	2
3	3.28 dd (J=10.5, 5.1 Hz)	3.28 dd (J=10.4, 5.0 Hz)
18	0.97 s	0.97 s
19	0.33 d (J=4.3 Hz)	0.33 d (J=4.2 Hz)
	0.56 d (J = 4.3 Hz)	0.56 d (J=4.2 Hz)
21	0.87 d (J = 6.5 Hz)	0.87 d (J=6.5 Hz)
23	5.70 ddd (J=15.7, 8.1, 5.9 Hz)	_
24	5.52 d (J=15.7 Hz)	4.27 t (J=6.5 Hz)
26	1.34 s	5.01 br s
27	1.34 s	1.74 br s
28	0.88 s	0.89 s
29	0.97 s	0.97 s
30	0.81 s	0.81 s
НОО	7.32 br s	7.78 br s

TABLE 1. ¹H-Nmr Chemical Shifts of **1** and **2** (CDCl₃).

1 and 2 ($CDCl_3$).			
Carbon	1	2	
1	32.0	32.0	
2	30.4	30.4	
3	78.9	78.9	
4	40.5	40.5	
5	47.1	47.1	
6	21.1	21.1	
7	26.0	26.0	
8	47.9	48.0	
9	20.0	20.0	
10	26.1	26.1	
11	26.5	26.5	
12	32.8	32.9	
13	45.4	45.3	
14	48.8	48.8	
15	35.6	35.6	
16	28.1	28.1	
17	52.1	52.2, 52.1	
18	18.1	18.0	
19	29.9	29.9	
20	36.3	36.0, 35.8	
21	18.4	18.3, 18.2	
22	39.4	32.0	
23	130.7	27.6, 27.3	
24	134.5	90.4, 90.2	
25	82.3	143.9, 143.7	
26	24.4°	114.5, 114.2	
27	24.3°	17.2, 16.9	
28	19.3	19.3	
29	25.4	25.4	
30	14.0	14.0	

TABLE 2. ¹³C-Nmr Chemical Shifts of 1 and 2 (CDCl₃).

^{*}Interchangeable signals.

hydroperoxy group vicinal to the terminal isopropenyl group of the side-chain. A NOESY spectrum showed an nOe correlation between the exchangeable proton and H-26 (δ 5.01). The ¹³C-nmr chemical shifts of C-24 bearing the hydroperoxide substituent (δ 90.4, 90.2) were similar to those reported for 24hydroperoxide-24-vinylcholesterol (6). Reduction of **2** with LiAlH₄ afforded **4**, which was identified by direct comparison (tlc and ¹H-nmr) with an authentic sample (3,4).

The known compounds, cycloartanone, cycloartenone, 24-methylenecycloartanone, cycloartanol, cycloartenol [**5**], 24-methylenecycloartanol, lanostenol, lanosterol, and 24-ethylcholest-4-en-3one were identified by comparison of their spectral data (eims, 1 H-nmr, 13 C-nmr) with those reported in the literature (7–12).

The origin of compounds 1 and 2perhaps may be understood as a naturally sensitized photooxygenation of cycloartenol [5] in the plant. The combination of molecular oxygen with olefins is a well-known phenomenon. The reaction involves the formation of an allylic hydroperoxide from an olefin by a process involving abstraction of an allylic proton along with migration of the carbon-carbon double bond (13,14). As there are two different types of allylic protons in 5, two different products were obtained and as photooxygenation is non-stereoselective, an epimeric mixture was obtained in the case of compound 2.

Other naturally occurring hydroperoxides have been reported from some plants (15,16) and tunicates (6) and their origin has been discussed, confirming that these compounds are not artifacts. A natural hydroperoxidation was also suggested to explain the biogenesis of some cycloartane derivatives (17). Taking into account that in this case the extraction was performed in darkness and the processing conditions were mild, it is apparent that this photooxygenation must take place in the plant itself. It is known that since ${}^{1}O_{2}$ is potentially damaging to chloroplast membrane lipids, proteins, and nucleic acids, protective systems are expected to exist in plants to quench these species (18), especially in plants like T. recurvata, with extreme exposure to air and sunlight due to growth on the upper parts of trees and on utility cables.

It is noteworthy that compounds **3** and **4**, which probably derive from **1** and **2**, have been reported from different plants (3,5), including *Tillandsia usneoides* (4). In all cases the extracts were obtained from dried plant material. Compounds **1** and **2** were probably not reported previously because they decompose during drying.

EXPERIMENTAL

GENERALEXPERIMENTAL PROCEDURES.—Mps are uncorrected. The Ft-ir spectra were recorded on a Nicolet Magna-IR 550 instrument. The uv spectra were taken on a Hewlett-Packard 8451 A diode-array spectrophotometer. Optical rotations were recorded on a Perkin-Elmer 141 polarimeter. Eims were taken on a Trio-2 quadrupole mass spectrometer (VG Biotech). Nmr spectra were recorded on a Bruker AC-200 instrument at 200.1 MHz for ¹H and 50.3 MHz for ¹³C in CDCl₃ with TMS as internal standard.

PLANT MATERIAL.—*Tillandsia recurvata* was collected at Huerta Grande, Córdoba, Argentina, in November 1993 and was identified by Dra. Rosa Subils. A voucher specimen is located at the herbarium of the Museo Botánico de Córdoba [CORD], Argentina.

EXTRACTION AND ISOLATION.-Fresh plant material (1 kg) was extracted at room temperature in the dark with EtOH and then with CH₂Cl₂. H₂O was added to the EtOH extract and the solution was extracted with hexane. The hexane layer was combined with the CH₂Cl₂ extract and taken to dryness. The oily residue (9.5 g) was fractionated by dry-column flash chromatography on Si gel using hexane and mixtures of hexane/ CH2Cl2 of increasing polarity. The fraction eluted with hexane-CH₂Cl₂ (1:1, 155 mg) was subjected to hplc (column, YMC C_{18} , 5 μ m, 22.5 × 2.5 cm; eluent, MeOH) affording cycloartanone (15 mg), cycloartenone (3 mg), and 24-methylenecycloartanone (2 mg). The fraction eluted with CH₂Cl₂ (1.34 g) was fractionated by dry-column flash chromatography on reversed-phase (C18) using mixtures of H₂O/MeOH and MeOH/CH₂Cl₂ of decreasing polarity. The fraction eluted with MeOH (545 mg) on hplc under the same conditions yielded cycloartanol (37 mg), lanostenol (42 mg), 24-methylenecycloartanol (15 mg), 24ethylcholest-4-en-3-one (20 mg), cycloartenol (21 mg), 24-hydroperoxycycloart-25-en-3β-ol [2] (7 mg), 25-hydroperoxycycloart-23-en-3β-ol [1] (4 mg), and lanosterol (26 mg), some of which were re-purified by prep. tlc using CH₂Cl₂-EtOAc (9:1).

25-Hydroperoxycycloart-23-en-3β-ol [1]. White needles from hexane/Et₂O; mp 127–128°, [α]²⁵D +32° (c=0.43, CHCl₃); uv λ max (CHCl₃) 244 (ϵ 330) nm; ir (KBr) ν max 3380, 3266, 2938, 2867, 1462, 1377, 1370 cm⁻¹; eims (20 eV) m/z [M]⁺ 458 (4), 440 (12), 424 (29), 409 (26), 393 (20), 365 (22), 342 (17), 339 (21), 315 (39), 297 (46), 255 (41), 175 (88), 121 (100); ¹H-nmr data, see Table 1; ¹³C-nmr data, see Table 2.

24-Hydroperoxycycloart-25-en-3 β -ol [2]. White solid from MeOH/H₂O; mp 112-115°, [α]²⁵D +39° (c=0.10, CHCl₃); uv λ max (CHCl₃) 242 (ϵ 288) nm; ir (KBr) ν max 3433, 3276, 2926, 2855, 1466, 1375 cm⁻¹; eims (20 eV) m/z [M]⁺ 458 (2), 440 (15), 424 (20), 409 (32), 381 (20), 315 (25), 297 (28), 255 (16), 175 (100); ¹H-nmr data, see Table 1; ¹³C-nmr data, see Table 2.

CONVERSION OF 1 TO 3 (OR 2 TO 4).—To a solution of 2 mg of compound 1 (or 2) in 0.5 ml anhydrous Et₂O, an excess of LiAlH₄ was added and the suspension was stirred for 2 h at room temperature. The reaction mixture was treated with MeOH; the salts were removed by filtration and the filtrate was evaporated to dryness. The crude mixture was chromatographed on Si gel (CH₂Cl₂-EtOAc, 9:1) affording 0.5 mg each of compounds 3 and 4, which were identified by comparison with authentic samples (3,4).

The ¹H- and ¹³C-nmr and eims spectra of cycloartanone, cycloartanol, and cycloartenol [**5**] were in full agreement with literature data (2,9,11,19) and with spectra of authentic samples. The ¹H- and ¹³C-nmr and eims spectra of 24-methylenecycloartanol, lanostenol, lanosterol, and 24-ethylcholest-4-en-3-one were similar to those reported in the literature (7,10,12). The ¹H-nmr and eims spectra of 24-methylenecycloartanone and cycloartenone also agreed with those reported in the literature (7,8,10).

ACKNOWLEDGMENTS

We thank UMYMFOR (CONICET-FCEN, UBA) for spectra, and CONICET, Argentina, and Fundación Antorchas for partial financial support. We are grateful to Dr. Juan C. Oberti and Dra. Rosa Subils for the identification of the plant and Dr. Jorge A. Palermo and M. Florencia Rodriguez Brasco for their help in the collection of the plant material.

LITERATURE CITED

- G.M. Cabrera, M. Gallo, and A.M. Seldes, *Phytochemistry*, **39**, 665 (1995).
- A. Milon, Y. Nakatani, J.-P. Kintzinger, and G. Ourisson, *Helv. Chim. Acta*, 72, 1 (1989).
- M. Della Greca, A. Fiorentino P. Monaco, and L. Previtera, *Phytochemistry*, **35**, 1017 (1994).
- C. Djerassi and R. McCrindle, J. Chem. Soc., 4034 (1962).
- V. Anjaneyulu, G. Sambasiva Rao, and J.D. Connolly, *Phytochemistry*, 24, 1610 (1985).
- M. Guyot, D. Davoust, and C. Belaud, *Tetrahedron Lett.*, 23, 1905 (1982).
- R.T. Aplin and G.M. Hornby, J. Chem. Soc. (B), 1078 (1966).
- P. Ramchandra, M. Basheermiya, G.L.D. Krupadanam, and G. Srimannarayana, J. Nat. Prod., 56, 1811 (1993).
- W. Kamisako, C. Honda, K. Suwa, and K. Isoi, Magn. Reson. Chem., 25, 683 (1987).

- J. De Pascual Teresa, J.G. Urones, I.S. Marcos, P. Basabe, M.J. Sexmero Cuadrado, and R. Fernandez Moro, *Phytochemistry*, 26, 1767 (1987).
- 11. R.B. Boar and C.R. Romer, *Phytochemistry*, **14**, 1143 (1975).
- G.T. Emmons, W.K. Wilson, and J. Schroepfer, Jr., Magn. Reson. Chem., 27, 1012 (1989).
- H.H. Wassermann and J.L. Ives, *Tetrahe*dron, 37, 1825 (1981).
- A. Nickon and J.F. Bagli, J. Am. Chem. Soc., 83, 1498 (1961).
- 15. R.W. Doskotch, F.S. El-Feraly, E.H.

Fairchild, and C.-T. Huang, J. Chem. Soc., Chem. Commun., 402 (1976).

- F.S. El-Feraly, Y.-M. Chan, E.H. Fairchild, and R.W. Doskotch, *Tetrahedron Lett.*, 1973 (1977).
- W. Herz, K. Watanabe, P. Kulanthaivel, and J.F. Blount, *Phytochemistry*, 24, 2645 (1985).
- J.P. Knox and A.D. Dodge, *Phytochemistry*, 24, 889 (1985).
- H.E. Audier, R. Beugelmann, and B.C. Das, *Tetrahedron Lett.*, 36, 4341 (1966).

Received 1 May 1995